

### **chapter 11 chemical reactions pdf**

Chemical equilibrium  $\hat{\in}$  when the forward rate of reaction is equal to the reverse rate of reaction.  $\hat{\in}$  chemical reactions at eqm are reversible.

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7 In a chemical reaction Atoms aren't created or destroyed. A reaction can be described several ways: 1. In a sentence (every item is a word) Copper reacts with chlorine to form copper (II) chloride. 2. In a word equation (some symbols used) Copper + chlorine  $\hat{\in}$  copper (II) chloride 8 Symbols in equations (page 323) the arrow separates the reactants from the

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